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List at least three characteristic properties of acids and three of bases.
ACIDS pH less than 7 Have a sour taste Change the color of many

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indicators Are corrosive (react with metals) Neutralize bases Conduct an electric current. BASES pH greater than 7 Have a bitter taste Change the color of many indicators Have a slippery feeling Neutralize acids Conduct an electric current.

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KEY - acid base study guide Acids are chemical compounds that release hydrogen ions (H^+) when placed in water. For example, when hydrogen chloride is placed in water, it releases its hydrogen ions and the solution

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 - Identify conjugate acid-base pairs in Brønsted-Lowry acids and bases.
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dissolve in water. To show how you can recognize strong and weak acids. This section introduces one way to define acids, called the Arrhenius definition.

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