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If the same thing happened in either experiment, CO<sub>2</sub> was produced by the experiment. In the first experiment, Mg dissolves in HCl to make aqueous MgCl<sub>2</sub> -- there should be no CO<sub>2</sub> liberated. In the...

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This lab is an excerpt from Heath Chemistry, a textbook of laboratory experiments. Information regarding the lab and detailed procedure are provided on the following pages. Note: Change Step 2 from g. O g of copper (II) chloride to approximately 3.0 g - 4.0 g

~~Lab 4B (student copy)~~

Help for Heath 19c experiment? I don't know how to make the table for question # 5. "From the results of the temperature at which a precipitate dissolved, make up a table showing the temperature, the ksp and the solubility."

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